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Description

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The present invention refers to new supported components of catalysts for the polymerization of olefins CH₂=CHR, in which R is an alkyl with 1—4 carbon etoms or an eryl, end of mixtures of such olefins with each other and/or with ethylene, and to the catalysts obtained from such components.

In earlier patent applications in the name of the Applicant there have been disclosed catalysts endowed with high activity and high stereospecificity, obtained from metallorganic Al compounds, silicon compounds containing Si—O—C, Si—OCOR or Si—NR₂ bonds and from a solid comprising a halogenated trainium compound end an electron-donor compound, both supported on an ective Mg helide, the relectron-donor compound being chosen among specific classes of estate.

JP-A-79 94,590 (C.A. 92,7198d) relates to a solid catalyst compound prepared from an electron donor—treated Mg-helide and a Ti halide optionally treated with an electron donor.

JP—A—116,777 (Jap. Pat. Gaz., 15—21 April 1981, page 7) is concerned with catalysts for the polymerization of alphaoletins, comprising TI components supported onto Mg-hallide, conventional 1s electron-donor compounds in the form of esters, ethers, amines end the like, organo-Al compounds and silicon compounds (silinane).

It has now been unexpectedly found that it is possible to prepare highly active and highly stemeospacific catalysts also by employing electron-donor compounds different from those described in the arier patent applications, provided that the employed electron-donor compound belongs to certain 20 classes of compound, as specified hereinafter, and is for at least 70% by mole stratcable from the solid catalyst component by reaction under standard conditions with Al-triethyl, and that the surface area of the product after extraction is not less than 20 m/ds.

The cetalysts of the invention comprise the product of the reaction between et leest the following components:

a) an Al-alkyl compound, preferably an Al-trialkyl or a compound containing two or more Al atoms linked to each other through oxygen or nitrogen atoms or through SO₄ or SO₃ groups;

 b) an electron-donor-compound (or Lewis base) which, under the standard conditions of meesurement as indicated hereinafter, is reactive towards MgCI₂ but which results not to be completely complexed with ALE₁, at the equivalent point of a potentiometric titration under standard conditions;

c) a solid comprising a Ti halide and an electron-donor compound both supported on a Mg halide, such electron-donor compound being selected from the classes of: ethers, ketones, lectones, electron-donor compounds containing N, P and/or S atoms, and from the following esters:

- 1) hydrocarbyl esters of linear saturated dicarboxylic acids containing from 2 to 5 C atoms;
 - 2) esters of unsaturated polycarboxylic acids, in which two carboxyl groups ere linked to vicinal, double bond-forming carbon atoms and in which the hydrocarbyl radical or radicals of the COOR groups ere linear saturated or unsaturated radicals or cycloaliphatic radicals with 1—20 C atoms or hydrocarbyl esters of unsaturated linear or branched polycarboxylic acids with 1—20 carbon atoms, in which the carboxy groups are not linked to vicinal double bond-forming arbon atoms:
 - hydrocarbyl esters of aromatic meta- and para-dicarboxylic acids and hydrocarbyl esters of aromatic polycarboxylic ecids containing more than two cerboxyl groups;
 - hydrocarbyl esters of aromatic hydroxy compounds containing the OH groups in mete- or para-position, and esters of aromatic hydroxy acids the OH groups of which are in meta- or para-position with respect to the carboxyl group;
 - 5) esters RCOOR' the hydrocarbyl groups R and R' of which, which can be the same or different, are linear saturated or unsaturated radicals or cycloalliphatic andicals having from 1 to 20 carbon atoms, or R is an aryl, allylaryl or cycloallyl with 5—20 carbon atoms and R' is a hydrocarbyl radical or a heterocyclic rina with 5—7 etoms in the ring;
 - 6) hydrocarbyl esters of polycarboxylic acids in which at least one cerboxyl group is linked to an aromatic ring and at least one other is linked to a carbon atom of an alliphatic group or to a cycloaliphatic ring or at least two carboxyl groups ere linked to an aromatic ring through an alkylene group;
 - 7) esters of aromatic polycarboxylic acids containing at least two non-condensed aromatic rings, each of which beer a carboxyl group:
 - 8) esters of cerbonic acid with glycols and carbonic ecid derivatives of formula RO—CO—OR' wherein R and R' are the same or different acyl groups with 1—20 carbon atoms;
 - 9) esters of polyols and of monohydroxy-phenols;
 - 10) hydrocarbyl esters of acetylenic acids; the electron-donor compound being extractable from the solid for at least 70% by mols with Al-triethyl under standard measurement conditions, and the surface area of the solid subjected to extraction being higher than 20 m²/g.
- Electron-donor compounds belonging to the ebove specified classes and suitable for the preparation of the catalyst components of the invention are: diseasmly either, discobutyl either, barcophomone, triphenty phosphite, phenyl propionate, di-n-butyl succinate, dils-butyl succinate, dil-n-butyl malonate, dilsyllymionate, di-n-butyl maleate, eithy and propyl trichlorocaetate, POCI, triphentylphosphine, and propyl triphentylphosphine, and propylphosphine, and propylphosphine

$(C_6H_5CO-O-CO-OC_2H_8)$,

dispointly latter of 0-phenylena disceric acid. It is understood that the above classes of esters include elso the aderivetives thereof as for instance the derivetives containing heliogen atoms or unseturated to hydrocarbyl radicals. Ethyl trichloroscetate and diethyl allyl malonate listed above are examples of such derivatives.

oerrozeros.

All the above compounds are extractable from the solid catalyst component for at least 70% by mols by reaction with Al-triethyl. The surface area of the solid effer the treatment with Al-triethyl is higher than 20 m²/g and in the case of very active catalysts higher than 100 m²/g end in general comprised between 100 20 and 300 m²/g.

In the case of weak electron-donors, es certein ethers, it has been found that the ether can be substantially removed from the cetalytic solid during the preparation if the latter comprises treatments with substantially removed from the cetalytic components fall within the scope of the present invention. It is understood that the above donor compounds, when is subjected to the extractability test with Art-riethy before the ebove mentioned treatments, comply with the

test.

It has been found that it is possible to obtain catalyst components according to the invention also by employing certain electron-donor compounds, such as alkyi, eyclosly() or eryl esters of aromatic monocarboxylic acids, which in the preparetion conditions hitherto known do not form catelyst 20 components satisfying the requirements of this invention. The components eccording to the invention and containing the above seters are prepared by methods which consist in hot westing with a halogeneted

containing the above sters are prepared by methods which consist in hot weshing with a halogeneted hydrocarbon solvent a co-ground mixture comprising the seter, the TI compound and the Mg helide. For TI halides and electron-donor compounds supported on e Mg dihalide are meent the compounds not extractable from component c) after treatment with boilling 1,2-dichloroethene for 2 hours?

25 (concentration of the solid in the suspension: 5% by weight). The compounds reactive towerds MgCl₂ ere those which under the standard conditions of the reaction

remain fixed on the Mg halide for at least 20% by mols.

Components e), b) end c) ere reacted with each other in whichever order; nevertheless, components a)

and b) are preferably pre-mixed before being contected with component c).

Component c) can be pre-mixed with component e) end/or with component b). The pre-mixing of al and b) is carried out at temperatures usually comprised between room temperature and the temperature.

employed in the polymerization.

The pre-reaction of c) and b) can be cerried out also et higher temperatures. Compound b) can elso be

The pre-reaction of c) and b) can be cerried out also et higher temperatures. Compound b) can elso b incorporated in and reected with component c) itself.

Component b) is reacted in a molar ratio with respect to the helogeneted Ti compound supported on component c) of at least 1, and in a molar ratio with respect to the Al-alkyl compound employed as component a) lower than 20 and preferebly comprised between 0.05 end 1. Ratios higher then 1 can be employed with compounds b) not complexing or only weekly complexing Al-triethyl elso under conditions promoting such complex formation.

In component c) the moler ratio between the Mg dihelide and the halogenated Ti compound supported on it is comprised between 1 end 500, end the moler ratio between the helogenated Ti compound end the electron-donn's but supported on the Mg dihelide is comprised between 0.1 and 50.

electron-donor norm supported on the my artists of the state of the st

variation of potential, i.e. a wave, in the discours in the case of amines like isoquinoline or esters like ethyl-p-toluate or ethylbenzoate, wherein the titration curve shows a wave at the equivalent point. The absence of a wave at the equivalent point indicates that electron-donor compound b) is present, at least in part, in not complexed form with Al-triethyl.

o Electron-donor compounds containing active hydrogen etoms reactive towerds Al-triethyl, i.e. cepeble of substitution reactions with Al-triethyl, ere prereacted with Al-triethyl, before subjecting the seme to the complexetion test. In the titration test the equivalent point is generally referred to the use of 1 mol of Al-triethyl per mol of donor.

Compounds b) can be used in mixture, in a wide renge, with donors forming complexes with Al-triethyl

55 when titrated according to the test set forth above. The titration of the mixtures containing significant amounts of the complexing donors, shows the potential veriation due to the presence of said complexing donors.

potential vertation due to the presented or sale completing deficiently using the above mixtures as However the polymerization performance of the catalysts obtained by using the above mixtures as component b) is not significantly reduced by the presence of the complexing donor.

The compound b) is selected in particular from compounds of general formule:

RmSiYnXp

wherein:

erem. R is an alkyi, alkenyi, aryi, arylalkyi or cycloalkyl radical with 1—20 carbon atoms; Y is en —OR', —OCOR' or —NR'₂ redical in which R', the same as or different from R, has the same meaning as R;

X is a halogen or hydrogen atom or a —OCOR" or —NR", group In which R", the seme as or different from R', has the seme meaning as R'; m, n, p ere numbers comprised:

m from 0 to 3, n from 1 to 4 and p from 0 to 1; m+n+p is equal to 4.

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Preferred silicon compounds are: phenyl alkoxy silanes, es phenyl triethoxy or phenyl trimethoxy silane, cliphenyl dimethoxy and diphenyl diadhoxy silane, monochlorophenyl diethoxy silane; alkyl alkoxy silanes, as ethyl triethoxy silanes.

In the catalysts according to the invention the silicon compound is present in combined for lor in the 10 soil product of the reaction between the various components forming the cetalyst in e molar ratio between the silicon compound and the halogenated Ti compound higher than 0.05 and in general comprised between 0.1 and 5.

Other sultable compounds b) are 2,2,6,6-tetramethylpiperidine, 2,2,5,5-tetramethylpyrrolidine, 2,2,6,6-tetramethylpiperidide-Al-diethyl, Al-dichloro-monophenoxy.

The Al-alkyl compounds of component a) include the Al-trialkyls, such as for instence AlEts, Al(i-Bu)s, Al(i-C₃H₃)s, AlEtsH, and compounds containing two or more Al atoms linked to each other through stero-atoms, such as:

$$\begin{split} &(\bar{C}_2H_0)_2A\bar{I}-\bar{O}-AI(C_2H_0)_2,\\ &(C_2H_0)_2A\bar{I}-N-AI(C_2H_0)_2,\\ &-\bar{C}_0H_0\\ &\bar{C}_0H_0\\ &\bar{C}_0H_0\\ &\bar{C}_2H_0)_2A\bar{I}-\bar{O}-\bar{S}-\bar{O}-AI(C_2H_0)_2\\ &\bar{C}_0H_0\\ &\bar{C}_0H_0$$

As indicated. Al-alkyl compounds in which Al etoms are linked through groups like SO_4 or SO_3 are also suitable.

The Al-alkyl compounds can be employed in admixture with Al-alkyl helides, as AIEt, Cl.

Component oil prepared according to various methods. One of these consists in co-grinding the Mg 35 hallde and the electron-donor compound until the product, after extraction with Al-tristly under standard conditions, shows a surface area higher than 20 m³/g, and in reacting subsequently the ground product with the TI compound.

Preparetions of this type ere described in British Patent No. 1,559,194.

Another method consists in reecting an adduct Mg helide/alcohol with a Ti compound in the presence 40 of the electron-donor compound. This method is described in BE—A—868,682.

According to another method, described in DE-A-3,022,738, the adduct Mg dihalide/alcohol is reacted in liquid form with the halogenated Ti compound and with the electron-donor compound.

Other methods are described in DE—A—2,224,029, US—A—4,220,554 and EP—A—15243.

Another method consists in co-grinding the Mg difialide, the halogenated TI compound and the 4s electron-donor compound until activation of the Mg difialide, in treating e suspension of the ground product in a halogeneted Mydrocarbons, such as 1,2-dichloresthane, hiorobenzene, methylene dichloride, hexachlorosthane, and in separating the solid from the Ilquid halogenated hydrocarbon. This method is particularly suitable for preparing catalyst components satisfying the extractability criterion of the present invention when es electron-donor compounds ere employed esters of arometic monocarboxylic acids, such as alkilv actives of benzoic acid.

The treatment is carried out et temperatures comprised between 40°C and the boiling point of the halogenated hydrocarbon for a time ranging in general from 1 to 4 hours.

According to another method, a porous carrier such as SiO₂ or Ai₂O₃ heving e low content of OH groups (preferably less than 1% by weight) is impregnated with a fleutid adduct Mg dihalide/alcohol; then ss the carrier is treated with an excess of TiCl₄ containing in solution the electron-donor compound according to the procedure described for example in DE—A—No. 3,022,738 or in BE—A—888,682.

In all the above methods the final product contains a Mg dihalde in the active form as defined hemiantars. For "active form of the Mg dihalde" is meen the Mg dihaldes showing in the X-ray powder spectrum of component c) a broadening of at least 30% of the most intense diffraction line appearing in the spowder spectrum of the corresponding dihalde having a surface area of 1 m/g, or is meent the Mg dihaldides showing a X-ray spectrum in which said most intense diffraction line is replaced by a halo with its intensity peak shifted with respect to the interplaner disease of the most Intense line of the most Intense lines.

Very active forms of Mg dihalides are those showing a X-ray powder spectrum in which the most intense diffrection line appearing in the spectrum of the corresponding hallde having a surface area of 1.55 m²/a has a decreesed intensity and is broedened to form a halo, or are those in which said most intense line

is replaced by e halo having its intensity peek shifted with respect to the interplaner distance of the most intense line.

In general, the surface area of the above forms is higher than $30-40 \text{ m}^2/g$ and in particular is comprised between 100 and $300 \text{ m}^2/g$.

Preferred Mg dihalides are MgCl₂ and MgBr₂. The water content of the dihalides in general is lower

than 1% by weight.

Other known methods for preparing a Mg dihalide in ective form or catalyst components containing Ti
and supported on a Mg dihalide, in which components the dihalide is present in ective form, are based on
the following reactions:

- reaction of a Grignard compound or of a compound MgR₂ (R is hydrocarbyl) or of complexes MgR₂Al-trialkyls with halogenating agents, such as AlX₃ or compounds AlRmXn (X is halogen, R is hydrocarbyl, m+n=3, SiCl₄ or HSiCl₅
- reaction of a Grignard compound with a silanol or a polysiloxane, H₂O or with an alcohol and subsequent reaction with a helogenating agent or with TiCl₄;
- subsequent reaction with a neiogenating agent or with TLU4;

 reaction of Mg with an alcohol and with a hydrogen halide, or of Mg with a hydrocarbyl halide and with
 an alcohol:
- reaction of MgO with Cl2 or AlCl3;
- reaction of MgX2 · nH2O (X is halogen) with a halogeneting agent or with TiCl4;
- 20 reaction of Mg mono- or dialcoholates or of Mg carboxylates with a helogeneting agent.

Titanium halides and halogen-alcoholates are in perticuler the TI tetrahelides, the TI trihalides and the TI trihalogen-alcoholates. Preferred compounds ere: TiCl_e: TiBr_e: 2.6-dimethylphenoxytrichlorottianium and trichloro-phenoxytianium.

5 The Ti trihalides are obtained by known methods, for instence by reduction of TiCl₄ with AI or with a metallorganic compound of AI or with hydrogen.

In the case of the Ti trihalides it can be expedient, to improve the performance of the catalysts, to carry out an oxidation, even partial, of the titanium during or efter the preparation of component c). To this

out an oxuserion, year partial, or its claims of the purpose halogens and lodine halides can be employed.

Preferred catelysts are those in which component c) is obtained from MgCl₂, TiCl₄ or Cl₃TiOC₄H₆ and seters of meleic or malonic acid or from dissoamyl other or benzophanone, and in which component b) is

phenyl- or ethyl-triethoxy-silane or diphenyl-dimethoxy-silane or diphenyl-diethoxy-silane.
Component a) is an Al-trialkyl, such as Al-triethyl or Al-trilsobutyl.
The preferred methods for the preparation of component o

35 BE—A—888,682, DE—A—2,924,029, US—A—4,220,554, EP—A—15243 or DE—A—3,022,738. Among the preferred methods for preparing component of there is also the co-grinding of MgCl₂, TiCl₄ and electron-donor and the treatment of the ground product with e helogenated hydrocerbon, as

and electron-donor and the treatment of the ground product with a helogenated hydrocerbon, as 1,2-dichlorosthane.

The catalysts according to the invention are employed to polymerize alpha-olefins by known methods, 40 that is by carrying out the polymerization in liquid phase, either in the presence or in the absence of an inert

hydrocarbon solvent, or in gas phase or also by combining, for instence, a polymerization step in liquid phase with a step in ges phase.

In general, the temperature is comprised between 40° and 160°C, but preferably between 60° and 90°C,

operating either et atmospharte pressure or at higher pressure.

As a molecular weight regulator can be employed hydrogen or another regulator of known type.

The catelysts are particularly suitable for the polymerization of propylene, butene-1, styrene and

4-methy-pentene-1. The catelysts can also be employed for polymerizing, by known methods, mixtures of

propylene and ethylene to produce modified polypropylenes having improved impact resistance at low

temperatures (the so called propylene/ethylene block copolymers) or to obtain crystalline random

copolymers of propylene with minor emounts of ethylene.

The test for the determination of the complexability of compound b) is carried out by using a potentiograph Metrohm mod. E 336 equipped with titration bench E 335, eutometic burette E 552, magnetic stirrer E 549 and titration cell EA 880. It is employed a combined electrode EA 281 (PW/Ag/AgC/KC/ 3M).

As titrating agent is employed a 0.5 M hexane solution of Al-triethyl which is added to a 0.5 M benzane solution of the compound under examination. It is operated at room temperature under nitrogen atmosphere. The test of reactivity of the electron-donor compound with MgCl_x is carried out under the following conditions.

Into a flask of 500 cc capacity are introduced, under nitrogen atmosphere, 2 g MgCl, (21 m Moles) suspanded in 200 cc toluene and 3.5 m Moles of the electron-donor under exemination. It is allowed to react at 25°C for 1 hour and the solid is filtered and weshed with 200 cc toluene and then with 200 cc n-heptane. The solid is Isolated, dried and enelyzed.

As MgCl₂ is employed the product obtained from MgCl₂ · 2.5 C₂H₆OH by reaction with Al-triethyl according to the following procedure: into a 3000 ce flast are introduced 2340 cc of a 0.83 M hexane solution of Al(C₂H₆), while maintaining the temperature below 10°C, 136 g MgCl₂ · 2.5 C₂H₆OH are added

little by little. After the eddition is completed, it is heated at 70°C for 4 hours; then the solid is filtered, washed repeatedly with n-heptane and dried under e vacuum of 0.2—0.5 Torr.

The surfece area of the obtained MgCl2 amounts to 618 m2/g and the pores volume to 0.532 cc/g.

5 Examples 1—16 and Comparetive Examples 1—4. Preparation of the solid catalyst component

Into a vibrating mill heving a total volume of 1 liter and containing 3 Kg steel balls of 16 mm diameter were introduced, under nitrogen stronghers, 50 g anhydrous MgCl₂, an electron-donor compound in such en emount as to have a molar ratio MgCl₂/monofunctionel electron-donor equal to 6 end a molar ratio MgCl₂/fincinctionel electron-donor equel to 12, end TiCl₂ in the molar ratio 1/1 with respect to the electron-donor compound. It was ground for 72 hours. 10g of the co-ground product were treated with 100 cc 12-dichloroethane for 2 hours at 80°C; it was filtered at the same temperature end efter several washings with n-heptane the solid cetalyst component was suspended in heptane.

15 Polymerization

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into a stainless steel autoclave of 3000 oc holding capacity, equipped with magnetic enchor stirrer and thermometer, heat stabilized at 60°C, into which propylene wes made to flow, was introduced a suspension consisting of 1000 oc enhydrous end deserated n-heptene, 5 millimoles aluminum triethyl, the desired amount of the silicon compound and the solid catalyst component.

Pydrogen was then introduced at a pressure of 0.2 atmospheres and it was heated rapidly at 70°C while simultaneously feeding in propylene up to a total pressure of 7 atmospheres. Seld pressure was to constent over the whole polymerization time by feeding in continuously the monomer. After 4 hours the polymerization was stopped and the polymer was isolated by filtration and dried. The quantity of polymer sisolated in the filtrate was isolated, weighed and summed to the polymer soluble in bolling n-heptane for the calculation of the isotacticity index (I.i.).

The results of the polymerization runs ere reported in the Teble.

In the Table ere also reported the results of the tests of extractability with Al-triathyl. The tests were carried out under the following standard conditions: into a 5 liber fleak are introduced at liber anhydrous and deserated n-heptane containing 20 m Moles AIC₂H₃. The temperature is raised to 70°C, about 1 g of solid catelyst component is introduced end the whole is allowed to recett under string for 15 minutes. Thereupon it is cooled rapidly at 0°C and is filtered, thus isolating a solid which is dried under vacuum at 25°C and analyzed.

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TABLE

[Solid catalyst component								
5		Electron donor (ED)	Extraction with AI(C ₂ H ₈) ₃							
	Example No.	Туре	% by weight (1)	% by weight (3)	Surface area (m²/g)					
10	1	Di-n-butyl maleate	6.2	0	186					
15	2	Di-n-butyl itaconate								
	3	Diethyl allylmalonate								
	4	Di-n-butyl malonate								
20	5	Diisobutyl succinate								
	6	Di-n-butyl succinate								
	7	Benzophenone								
25	8	Ethyl trichloroacetate	9.2	0.54	186					
	9	Ethyl benzoate	8.8	0.8	163					
30	10	Diisoamyl ether (2)	1	0	163					
	11	a-Methyl-a-phenyl-y- butyrolactone	7.3	0.9	90					
35	12	POCI ₃								
	13	Triphenylphosphine								
	14	Ethyl acetate	4.2	0	229					
40		0 0								
	15	C ₆ H ₆ —Ü—O—Ü—OC₂H ₆	9.4	0.3	174					
45 50	16	Diisobutyl o.phenylene diacetate	6.7	0.9	165					
	1 compar.	2-Ethylhexylterephtalate	9.2	0.26	1					
	2 compar.	Phenothiazine	7.7	3.6						
	3 compar.	Diisobutyl O,O-dlphenyldi- carboxylate	6.6	2.4						
55	4 compar.	Isobutyl m-acetoxybenzoate	8.8	4.2						

0 086 473 TABLE (continued)

		Polymerization					
5	Example No.	Silicon compound	Al/Si molar ratio	Yield g PP/g catalyst component	i.l. (%)	ηin (dl/g)	
10	1	(C ₆ H ₅) ₂ Si(OCH ₃) ₂	10	8600	95.1	1.5	
10	2	do	10	4600	93.6	1.4	
	3	do	10	8500	97.3	1.6	
15	4	· do	10	5000	91.2	1.3	
	5	do	10	4000	90	1.2	
20	6	do	10	3600	91.5	1.3	
20	7	do	3.3	3600	93.8	1.2	
	8	(C ₆ H ₅)Si(OC ₂ H ₅) ₃	3.3	5600	92.2	1.1	
25	9	do	10	4500	95.5	1.1	
	10	do	5	5200	91	1.3	
30	11	do	5	3600	91.8	1.2	
30	12	do	5	4000	92	1.5	
	13	do	5	3600	90.5	1.2	
35	14	do	10	3200	96.6	1.2	
	15	do	20	5500	91.3	1.5	
40	16	do	20	3500	90.7	1.2	
	1 compar.	do	10	2300	88	1.1	
45	2 compar.	do	10	1300	71.8	1.2	
	3 compar.	do	10	trace	_	-	
50	4 compar.	do	10	550	84.4	1.2	

⁽¹⁾ The percent amount of ED is referred to the solid after washing with 1,2-dichloroethane.

Claims

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A solid component to be employed, in combination with Al-alkyl compounds and with electron-donor compounds which results not to be completely complexed with Al-triethyl at the squivalent point of a potentiometric titration under standard conditions but reactive towards anhydrous Mg chloride, to form catalysts for the polymerization of alpha-olefins CH₂-CHR, said component comprising a titanium so compound having at least a Ti-halogen bond and an electron-donor compound both supported on an

⁽²⁾ The discomyl ether content of the ground product, before washing with 1,2-dichloroethane, was 17%

⁽³⁾ Percent of ED remaining on the solid of the extraction with Al(C₂H₅)₃.

anhydrous Mg dihelide, the electron-donor compound being selected from the group consisting of electron-donor compounds conteining N, P and/or S atoms, ethers, ketones, lectones end the following classes of seters:

- 5 1) hydrocarbyl esters of linear saturated dicarboxylic acids having from 2 to 5 carbon atoms;
 - 2) esters of unsetureted polycerboxylic acids in which two cerboxyl groups ere linked to vicinal, double bond forming cerbon etoms end in which the hydrocerbyl radical or radicals of the COOR groups are linear saturated or unsaturated are cyclatiphatic radicals with 1—20 cerbon atoms or hydrocarbyl asters of unsaturated linear or branched polycarboxylic acids with 1—20 cerbon atoms in which the carboxy groups ere not linked to vicinal double bond-forming carbon atoms;
 - hydrocarbyl esters of aromatic meta- and para-dicarboxylic acids and hydrocarbyl esters of polycarboxylic aromatic ecids containing more then two carboxyl groups;
 - 4) hydrocarbyl esters of aromatic hydroxy compounds containing the OH groups in meta- or para-position, and esters of aromatic hydroxy acids the OH groups of which are in meta- or para-position with respect to the carboxyl group;
 - 5) esters RCOOR' in which the hydrocarbyl groups R end R', which can be the same or different, are linear saturated or unseturated radicals or cycloaliphatic radicals having from 1 to 20 carbon atoms, or R is an aryl, alkylaryl or cycloalkyl with 5—20 cerbon etoms end R' is e hydrocarbyl redical or e heterocyclic ring with 5—7 atoms in the ring;
- WILLD For a doors in the limit of the sound of the sou
 - least two carboxy; groups are linked to an atomotic ring discognized season of the state of a carboxyl group;

 Nich bear a carboxyl group;
- without user a variously group.

 28 Sesters of carbonic acid with glycols; and carbonic acid derivatives of formula RO—CO—OR' wherein R and R' ere the xame or different acyl group with 1—20 carbon atoms;

 9) esters of polyols and of monohydroxy-phenols;
 - 10) hydrocarbyl esters of acetylenic acids; the electron-donor compound being extractable from the solid for at least 70% by mols with Al-triethyl under stendard measurement conditions, and the surface area of the solid subjected to extraction being higher than 20 m²g.
 - 2. A cetelyst component according to claim 1, In which the electron-donor compound is selected from drag roup consisting of discounyl either, benzohenone, triphenyl phosphile, dri-butyl mionete, clientyl allylmoinete, dri-butyl maleate, at 11 microholdre or triphoro-eroty-tienlum.
- ally/metionete, din-n-buryl maleate, ethyl trichlorocestate, propyl trichlorocestere, PCOL₂, In-reasonation did-p-toluset, and the Ti compound is Ti etrachloride or trichloro-erploxy-tientlum: 3. Cetalysts for the polymerization of alpha-olefins, comprising the product of the reaction of: al an Al-alkyl compound, in particular an Al-triakly or an Al-alkyl compound, in particular an Al-triakly or an Al-alkyl compound.
 - etoms linked to eech other through oxygen or nitrogen atoms or through SO₄ or SO₃ groups; b) an electron-donor compound reactive, under standard conditions, towards MgCl₂ but not
- 40 completely complexed with Al-triethyl;
 - c) the catalyst component of the preceding claims 1 and 2.
 4. Catalysts according to claim 3, in which component b) is selected from silicon compounds of general formula:

RmSiYnXp

wherein

- R is an alkyl, alkenyl, aryl, erylalkyl or cycloelkyl redical with 1-20 carbon atoms;
- R is an alklyl, alkenyl, aryl, erylarkyl or cyclodikyl hedical with 1250 some R, has the same as or different from R, has the same meaning as R;
- X is a halogen or hydrogen atom or a -0COR" or -NR"₂ group in which R", the same as or different from R', has the seme meaning as R'; m, n, p are numbers comprised:
 - m from 0 to 3, n from 1 to 4 and p from 0 to 1; m+n+p is equel to 4. 5. Use of the catalysts according to anyone of cleims 3 and 4 for the polymerization of alpha-olefins.

55 Patentensprüche

1. Eine feste Komponente zur Verwendung in Kombination mit Al-alkylverbindungen und mit Elektronen-Donorverbindungen, die mit Al-triethyl bei einem Gleichegweichtspunkt einer potentiometrischen Titretion unter Standardbedingungen nicht vollständig komplexiert wird, jadoch gegenüber 60 wesserfreiem Keine bei der Weine der We

- (1) Hydrocarbylester lineerer, gesättigter Dicarbonsäuren mit 2 bis 5 C-Atomen;
- (2) Ester ungesättigter Polycarbonsäuren, worin zwel Carboxylgruppen en vicinale, Doppeblindungen bildende Kohlenstoffstoren gebunden sind und worin der Hydrocarbytrest dodr Reste der CODR-Gruppen lineare, gesättigte oder ungesättigte Reste sind oder cyclosliphatische Reste mit 1 bis 20 C-Atomen, oder Hydrocorbylester ungesättigter, linearer oder vorzweigter Polycarbonsäuren mit 1 bis 20 Kohlenstoffstomen, worin die Carboxygruppen nicht an vicinale, Doppelbindungen bildende Kohlensötfstomen abunden sind:
- (3) Hydrocarbylester erometischer meta- und pere-Dicarbonsäuren und Hydrocarbylester erometischer Polycarbonsäuren, enthaltend mehr als zwei Carboxylgruppen;
- (4) Hydrocarbylester aromatischer Hydroxyverbindungen, enthaltend die OH-Gruppen in meta- oder para-Stellung, und Ester eromatischer Hydroxysäuren, deren OH-Gruppen in meta- oder pera-Stellung in bezug auf die Carboxydrupone sind.
- (5) Ester RCOOR*, wobel die Hydrocarbylgruppen R und R*, die gleich oder verschieden sein k\u00f6nnen, lineare, gestitigte oder unges\u00e4titigte Reste oder cycloeliphetische Reste mit 1 bis 20 K\u00f6henstoffsatomen sind, oder R lat ein Aryl, Alkylaryl oder Cycloelkyl mit \u00e5 bis 20 Kohlenstoffstomen und R* ist ein Hydrocarbylerst oder ein heterocyclischer Ring mit 5 bis 7 Atomen im Ring;
- (6) Hydrocarbylester von Polycarbonsäuren, worin wenigstens eine Carboxylgruppe an einen aromatischen Ring gebunden ist und wenigstens eine endere an ein Kohlenstoffatom einer alliphatischen Gruppe oder an einen cycloeliphetischen Ring gebunden ist oder wenigstens zwei Carboxylgruppen sind an einen aromatischen Ring über eine Alklengruppe gebunden;
- Sterr aromatischer Polycarbonsäuren, enthaltend wenigstens zwei nicht-kondensierte, aromatische Ringe, wobei jeder devon eine Cerboxylgruppe trägt;
- (8) Ester von Kohlensäure mit Glykolen und Kohlensäure-Derivate der Formel RO—CO—OR', worin R und R' die gleichen oder verschiedene Acylgruppen mit 1 bis 20 Kohlenstoffatomen sind;
- 5 (9) Ester von Polyolen und von Monohydroxyphenolen;
- (10) Hydrocarbylester acetylenischer S\u00e4uren; wobel die Elektronen-Donorverbindung eus dem Faststoff zu mindestens 70 Mol-\u00e8 mit Al-triethyl unter Standard-Me\u00dfbedingungen extrahlerbar Ist und der Oberf\u00e4\u00e5hehenbereich des Feststoffs, der der Extraktion unterworfen wird, h\u00f6her als 20 m\u00e7g ist.
- 2. Eine Katalysatorkomponente gemäß Anspruch 1, worin die Elektronen-Donorverbindung eusgewählt ist aus der Gruppe, bestehend aus Diisoamylieten, Benzophenon, Triphenrylphosphit, Di-n-butylmalonat, Diethylallylmalonat, Di-n-butylmeiset, Ethyltrichloracetat, Propylirtichloracetat, PCDI₃, 1,6-Hexandiol-di-p-tolluat, und wobei die Ti-Verbindung Titantetrachlorid oder Trichlioraryloxytiten ist. 3. Katalysatoren zur Polymerisation von «Olefinen, umfassend das Produkt der Reaktion von
- Katalysatoren zur Polymerisation von α-Olefinen, umfassend das Produkt der Reaktion von (a) einer Aleikyverbindung, insbesondere eine Arträlkyl- oder eine Al-Reikyverbindung mit zwei oder mehr Aluminiumatomen, die miteinander über Sauerstoff- oder Stickstoffatome oder über SO₄- oder SO₄-Gruppen verbunden sind;
- (b) einer Elektronen-Donorverbindung, die unter Standardbedingungen gegenüber MgCl₂ reaktiv ist, iedoch mit Al-triethyl nicht vollständig komplexiert wird:
 - (c) der Katalysatorkomponente der vorstehenden Ansprüche 1 und 2.
 - Katalysetbren gemäß Anspruch 3, worln die Komponente (b) ausgewählt ist aus Sillciumverbindungen der allgemeinen Formel

RmSiYnXp

45 worin:

- R ein Alkyl-, Alkenyl-, Aryl-, Arylalkyl- oder Cycloalkylrest mit 1 bis 20 Kohlenstoffatomen ist; Y ein Rest —OR', —OCOR' oder —NR', sit, worin R' gleich oder von R verschieden ist und die gleiche Bedeutung wie R hat;
- X ein Helogen- oder Wasserstoffatom oder eine Gruppe —OCOR" oder —NR"₂ ist, wobei R" gleich 50 oder verschieden von R' ist und die gleiche Bedeutung wie R' hat;
 - m, n und p Zahlen sind in dem Bereich:
 - m von 0 bis 3, n von 1 bis 4 und p von 0 bis 1; m+n+=4.
 - 5. Verwendung der Katalysatoren gemäß irgendeinem der Ansprüche 3 und 4 zur Polymerisation von α-Olefinen.

Revendications

contenant des atomes de N, de P et/ou de S, les éthers, les cètones, lactones et les classes suivantes d'esters:

- 1) des esters hydrocarbonés d'acides polycarboxyliques linéaires seturés contenent de 2 à 5 atomes de carbone:
- 2) des esters d'acides polycarboxyliques insaturés dens lesquels deux groupes carboxyles sont liés à des atomes de carbone formateurs de double liaison vicinaux et dans lesquels le ou les radicaux hydrocarbonés des groupes COOR sont des radicaux linéaires saturés ou insaturés ou des radicaux cycloellphatiques comprenant de 1 à 20 atomes de carbone, ou des esters hydrocerbonés d'ecides polycerboxyliques linéaires ou ramifiés inseturés comprenant de 1 à 20 etomes de carbone, dans lesquels les groupes carboxy ne sont pes liés à des atomes de carbone formateurs de double lieison vicineux;
- 3) des esters hydrocarbonés d'acides méta- et para-dicarboxyliques aromatiques et d'esters hydrocarbonés d'acides polycarboxyliques arometiques contenant plus de deux groupes carboxyles;
- 4) des esters hydrocerbonés de composés hydroxy aromatiques contenant des groupes OH en position métal- ou para-, et des esters d'acides hydroxy aromatiques dont les groupes OH sont en position méta- ou para- par rapport ou groupe carboxyle;
- 5) des esters RCOOR' dont les groupes hydrocerbones R et R', qui peuvent être identiques ou différents, sont des redicaux saturés ou insaturés linéeires ou des radicaux cyclo-aliphatiques comprenant 1 à 20 atomes de carbone, ou R est un redical aryle, alkylaryle ou cycloelkyle comprenent de 5 à 20 etomes de carbone et R' est un radicel hydrocarboné ou un noyau hétérocyclique comprenant 5 à 7 etomes dans le cycle:
- 6) des esters hydrocarbonés d'acides polycarboxyliques dans lesquels d'au moins l'un des groupes carboxyliques est relié à un noyau aromatique et au moins un autre est relié à un atome de carbone d'un groupe eliphatique ou à un noyeu cyclo-eliphatique ou au moins deux groupes carboxyles sont liés à un noyau aromatique par un groupe alkylène;
- 7) des esters d'acides polycarboxyliques eromatiques contenant eu moins deux noyeux arometiques non condensés, chacun d'entre eux portant un groupe carboxylé;
- 8) des esters d'acide carbonique avec des glycols et des dérivés d'acide carbonique de formule RO-CO-OR' dens lequel R et R' sont des groupes acyles identiques ou différents comprenent 1 à 20 atomes de carbone;
- 9) des esters de polyols et de monohydroxy-phénols;
- 10) des esters hydrocarbonés d'acides acétyléniques; le composé donneur d'électrons étant susceptible d'être extrait du solide par eu moins 70% en moles à l'aide d'un triéthylaluminium dans les conditions de mesure standards, et l'aire superficielle du solide soumis à l'extraction étant supérieure à 20 m²/g.
- Un constituent catalytique selon la revendication 1, dans lequel le composé donneur d'électrons est choisi dans le groupe comprenant l'éther diisoamylique, le benzophénone, le phosphite de triphényle, le malonate de di-n-butyle, l'allyl-malonate de diéthyle, le maléate de di-n-butyle, le trichloroecétate d'éthyle, le trichloroecétete de propyle, POCl₃, le 1,6-hexanediol -i-p-toluete, et le composé de titane consiste en tétrachlorure de titane ou trichloro-aryloxy-titane.
 - 3. Catalyseurs pour la polymérisation d'alphe-oléfines, comprenant le produit de la réaction de: a) un compose Al-alkyle, de préférence un Al-trialkyle ou un Al-alkyle contenant deux ou plusieurs
- atomes d'aluminium, liés entre eux par des atomes d'oxygène ou d'azote ou per des groupes SO4 ou SO3; b) un composé donneur d'électrons réactif, dens les conditions standards, evec MgCl₂ mais n'étant pas
 - complètement complexé par le triéthyleiuminium; c) le constituant catalytique des revendications 1 et 2 précédentes.
- - 4. Catalyseurs selon la revendication 3, dans lesquels le constituant b) est choisi parmi les composés de silicium de formule générale:

RmSiYnXn

dans laquelle:

R est un radical alkyle, alkényle, aryle, arylalkyle ou cycloalkyle comprenant 1 à 20 atomes de carbone; y est un radical -OR', -OCOR' ou -NR', dans lequel R', identique ou différent de R, a la même 55 signification que R;

X est un atome d'hydrogène ou halogène ou un groupe -OCOR" ou -NR"2 dans lesquels R", identique ou différent de R', a la même signification que R';

m, n, p sont des nombres compris:

m entre 0 et 3, n entre 1 et 4 et p entre 0 et 1;

60 m+n+p étant égal à 4.

5. Utilisation des catalyseurs selon l'une quelconque des revendications 3 et 4 pour la polymérisation des elphaoléfines.

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